

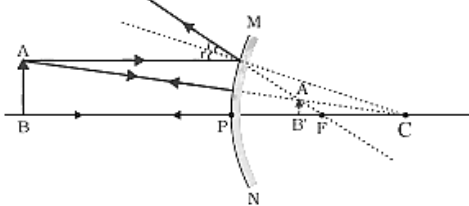
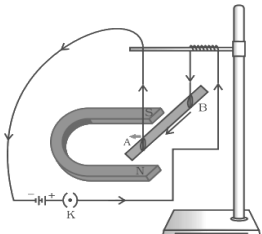
MARKING SCHEME
SCIENCE (Subject Code-086)
(PAPER CODE: 31/1/2) (10-01-86K)

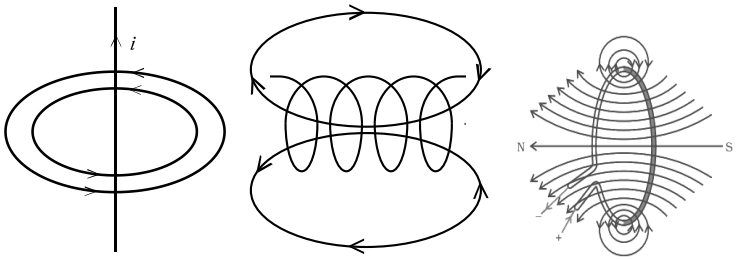
Q.No.	EXPECTED OUTCOMES/VALUE POINTS	Marks	Total Marks																						
SECTION – A (Biology)																									
1.	(C)/ (i), (ii) and (iv)	1	1																						
2.	(B) / Stomata	1	1																						
3.	(C) /Vegetative propagation	1	1																						
4.	(C) / Polythene bag, rubber band, ball pen	1	1																						
5.	(B) / Cerebellum	1	1																						
6.	(D) / they reproduce asexually	1	1																						
7.	(B) / Trypsin digests proteins and lipase digests emulsified fats	1	1																						
8.	(C) / Assertion (A) is true, but Reason (R) is false.	1	1																						
9.	(B) / Both, A and R are true, but Reason (R) is not the correct explanation of Assertion (A).	1	1																						
10.	<table border="1" style="width: 100%; border-collapse: collapse;"> <tr> <td style="text-align: center; width: 50%;">(a)</td> <td style="width: 50%;"></td> </tr> <tr> <td style="text-align: center;">Chewing of food</td> <td style="text-align: center;">Salivation on sight of food</td> </tr> <tr> <td>i) It is a voluntary action.</td> <td>i) It is a reflex/involuntary action.</td> </tr> <tr> <td>ii) It is controlled by forebrain.</td> <td>ii)It is controlled by medulla in the hind brain.</td> </tr> <tr> <td colspan="2" style="text-align: center;">(any other suitable difference)</td> </tr> <tr> <td colspan="2" style="text-align: center;">OR</td> </tr> <tr> <td style="text-align: center;">(b)</td> <td></td> </tr> <tr> <td style="text-align: center;">Pollination</td> <td style="text-align: center;">Fertilization</td> </tr> <tr> <td>i) It is the transfer of pollen grains from anther to the suitable stigma.</td> <td>i) It is the fusion of male gamete with the female gamete.</td> </tr> <tr> <td>ii) It occurs in plants.</td> <td>ii)It occurs in both plants and animals.</td> </tr> <tr> <td colspan="2" style="text-align: center;">(any other suitable difference)</td> </tr> </table>	(a)		Chewing of food	Salivation on sight of food	i) It is a voluntary action.	i) It is a reflex/involuntary action.	ii) It is controlled by forebrain.	ii)It is controlled by medulla in the hind brain.	(any other suitable difference)		OR		(b)		Pollination	Fertilization	i) It is the transfer of pollen grains from anther to the suitable stigma.	i) It is the fusion of male gamete with the female gamete.	ii) It occurs in plants.	ii)It occurs in both plants and animals.	(any other suitable difference)		1 1 1 1	2
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11.	Squirrel shows fight or flight response -Release of adrenaline, heart beats faster, more supply of oxygen to muscles (limb muscles), reduced blood supply to digestive system and skin, contraction of rib muscles and diaphragm, increased breathing rate.	2	2																						
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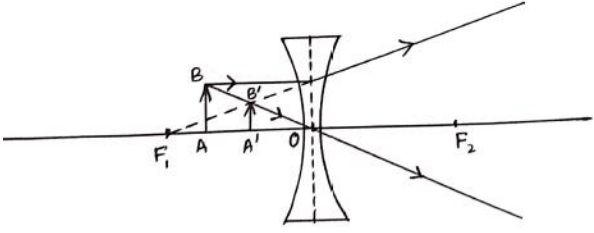
	OR		
	(c) (ii) Mendel's observations: <ul style="list-style-type: none"> All plants of F₁ progeny were tall. No medium/ no short height plants observed in F₁ progeny. F₁ progeny resembled one parent only. <p style="text-align: right;">(any two observations)</p>	1+1	4
16.	(a) <p>(i) When spores land on a substance and get adequate moisture and temperature, it will develop into new <i>Rhizopus</i>.</p> <p>(ii) New plants grow from the buds located in the notches of the leaf.</p> <p>(iii) The pollen tube will not be formed. / No fertilisation will take place.</p> <p>(iv) Fertilization /Pregnancy will be prevented.</p> <p>(v) Each fragment or piece grows into a new individual organism.</p> <p style="text-align: center;">OR</p> <p>(b) <p>(i) Most of these bacteria would die, but the few variants resistant to heat would survive and grow further.</p> <p>(ii) Fertilization occurs to form a zygote.</p> <p>(iii) Cross pollination may occur leading to fruit formation. / No fertilization. / No fruit formation.</p> <p>(iv) If the egg is not fertilised, the thick and spongy lining of the uterus breaks and comes out through vagina as blood and mucus, known as menstruation. / Menstruation will take place.</p> <p>(v) The seed will develop into a seedling. / Germination will take place.</p> </p>	1	
		1	
		1	
		1	
		1	
		1	
		1	
		1	5
SECTION – B			
(Chemistry)			
17.	(B) / Calcium	1	1
18.	(A) / tomato, curd, ant-sting	1	1
19.	(A) / Pb	1	1
20.	(B) / -COOH	1	1
21.	(D) / NO ₂ and O ₂	1	1
22.	(B) / Vanilla essence	1	1
23.	(C) / Both, (i) and (ii) are double displacement reactions and precipitation reactions.	1	1
24.	(A) / Both, Assertion(A) and Reason (R) are true and Reason (R) is the correct explanation of the Assertion (A).	1	1
25.	<ul style="list-style-type: none"> Tooth decay starts when pH of mouth is lower than 5.5. It can be prevented by cleaning the mouth after eating. / By using toothpastes as they are basic in nature. <p style="text-align: right;">(any other preventive method)</p>	1	
		1	2

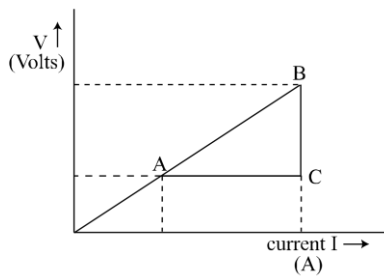
26.	<p>(a)</p> <ul style="list-style-type: none"> When electricity is passed through brine, it decomposes to form sodium hydroxide (alkali) and chlorine, hence this process is called chlor-alkali process. $2\text{NaCl}(\text{aq}) + 2\text{H}_2\text{O}(\text{l}) \xrightarrow{\text{Electricity}} 2\text{NaOH}(\text{aq}) + \text{H}_2(\text{g}) + \text{Cl}_2(\text{g})$ <ul style="list-style-type: none"> At anode: Cl_2 At cathode: H_2 <p style="text-align: center;">OR</p> <p>(b)</p> <p>(i) $\text{NaCl} + \text{H}_2\text{O} + \text{NH}_3 + \text{CO}_2 \rightarrow \text{NaHCO}_3 + \text{NH}_4\text{Cl}$</p> <p>(ii) $\text{Ca}(\text{OH})_2 + \text{Cl}_2 \rightarrow \text{CaOCl}_2 + \text{H}_2\text{O}$</p> <p style="text-align: center;">/</p> $2\text{Ca}(\text{OH})_2 + 2\text{Cl}_2 \rightarrow \text{Ca}(\text{ClO})_2 + \text{CaCl}_2 + 2\text{H}_2\text{O}$ <p>(iii) $\text{CaSO}_4 \cdot 2\text{H}_2\text{O} \xrightarrow{373\text{K}} \text{CaSO}_4 \cdot \frac{1}{2}\text{H}_2\text{O} + 1\frac{1}{2}\text{H}_2\text{O}$ (deduct $\frac{1}{2}$ mark for no / incorrect balancing)</p>	<p>1</p> <p>1</p> <p>$\frac{1}{2}$</p> <p>$\frac{1}{2}$</p> <p>1</p> <p>1</p> <p>1</p> <p>3</p>	
27.	<p>(a) $\text{CaO}(\text{s}) + \text{H}_2\text{O}(\text{l}) \rightarrow \text{Ca}(\text{OH})_2 + \text{Heat}$</p> <p>(b) $\text{CH}_4(\text{g}) + 2\text{O}_2(\text{g}) \rightarrow \text{CO}_2(\text{g}) + 2\text{H}_2\text{O}(\text{g})$</p> <p>(c) $2\text{FeSO}_4(\text{s}) \xrightarrow{\text{Heat}} \text{Fe}_2\text{O}_3(\text{s}) + \text{SO}_2(\text{g}) + \text{SO}_3(\text{g})$ (deduct $\frac{1}{2}$ mark for no / incorrect balancing)</p>	<p>1</p> <p>1</p> <p>1</p> <p>3</p>	
28.	<p>(a) Because it is easier to obtain metal from its oxide. / Because it is easier to reduce metal oxide to metal</p> <p>(b)</p> $\text{Fe}_2\text{O}_3(\text{s}) + 2\text{Al}(\text{s}) \rightarrow 2\text{Fe}(\text{l}) + \text{Al}_2\text{O}_3(\text{s})$ $3\text{MnO}_2(\text{s}) + 4\text{Al}(\text{s}) \rightarrow 2\text{Mn}(\text{l}) + 2\text{Al}_2\text{O}_3(\text{s})$ <p style="text-align: center;">(balancing is optional) (any one equation)</p> <p>(c) (i) $2\text{Cu}_2\text{S} + 3\text{O}_2(\text{g}) \xrightarrow{\text{Heat}} 2\text{Cu}_2\text{O}(\text{s}) + 2\text{SO}_2(\text{g})$</p> $2\text{Cu}_2\text{O} + \text{Cu}_2\text{S} \xrightarrow{\text{Heat}} 6\text{Cu}(\text{s}) + \text{SO}_2(\text{g})$ <p style="text-align: center;">OR</p> <p>(c)(ii) (I) Because highly reactive metals have more affinity for oxygen than carbon.</p> <p>(II) Because of its low melting point.</p>	<p>1</p> <p>1</p> <p>1</p> <p>1</p> <p>1</p> <p>1</p> <p>1</p> <p>4</p>	
29.	<p>(a) (i)</p> <p>(I) They do not have charged particles /ions.</p> <p>(II) Soaps react with calcium and magnesium salts present in</p>	<p>1</p> <p>1</p>	

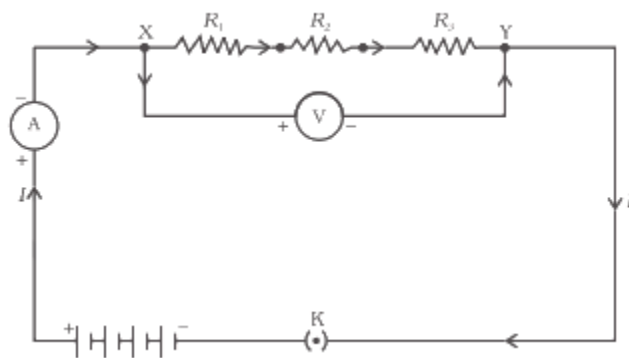
	hard water and form insoluble substances called Scum.		
	(III) C-C bonds are strong and stable whereas Si-Si bonds are relatively weak.	1	
	(ii) (I) $\text{CH}_3\text{CH}_2\text{OH} \xrightarrow{\text{acidified K}_2\text{Cr}_2\text{O}_7 + \text{heat}} \text{CH}_3\text{COOH}$	1	
	(II) $\text{CH}_2=\text{CH}_2 + \text{H}_2 \xrightarrow{\text{Ni}} \text{CH}_3-\text{CH}_3$	1	
	OR		
	(b) (i) X - CH_3COOH /ethanoic acid /acetic acid	½	
	Y - $\text{CH}_3\text{COOC}_2\text{H}_5$ / $\text{CH}_3\text{COOCH}_2\text{CH}_3$ / ester/ ethyl ethanoate	½	
	Z - CH_3COONa /sodium ethanoate/sodium acetate	½	
	(ii) Catalyst/ dehydrating agent	½	
	(iii)		
	•		
	$\text{CH}_3-\text{COOH} + \text{CH}_3-\text{CH}_2\text{OH} \xrightleftharpoons{\text{Acid}} \text{CH}_3-\overset{\text{O}}{\parallel}{\text{C}}-\text{O}-\text{CH}_2-\text{CH}_3 + \text{H}_2\text{O}$	1	
	• Esterification reaction	½	
	• $\text{CH}_3\text{COOC}_2\text{H}_5 + \text{NaOH} \rightarrow \text{CH}_3\text{COONa} + \text{C}_2\text{H}_5\text{OH}$	1	
	• Saponification reaction/ De-esterification reaction	½	5
SECTION – C			
(Physics)			
30.	(A) / Ciliary muscles of your eye contract and the eye lens become thick	1	1
31.	(C) / - 30 cm and + 30 cm from lens	1	1
32.	(C) / Assertion (A) is true, but Reason (R) is false.	1	1
33.	a) $r = 0.01 \text{ cm} = 1 \times 10^{-4} \text{ m}$ $l = 1 \text{ cm} = 0.01 \text{ m}$ $R = \rho \frac{l}{A}$ $\rho = \frac{RA}{l} = \frac{R \times \pi r^2}{l}$ $\rho = \frac{7 \times 22 \times 10^{-8}}{7 \times 0.01}$ $\rho = 22 \times 10^{-8} \times 10^2$ $\rho = 22 \times 10^{-6} \Omega\text{m} = 2.2 \times 10^{-5} \Omega\text{m}$	½	
		1	
		½	
	OR		

	<p>(b) Resistance of electric heater $R = \frac{V}{I}$</p> $R = \frac{220}{11}$ $R = 20 \Omega$ $P = \frac{V^2}{R}$ $P = \frac{200 \times 200}{20}$ $P = 2000W / 2 kW$	$\frac{1}{2}$ $\frac{1}{2}$ $\frac{1}{2}$ $\frac{1}{2}$	 2
34.	<ul style="list-style-type: none"> • Convex mirror / Diverging mirror •  <p>($\frac{1}{2}$ mark to be deducted for not showing the direction of light)</p>	1 1	 2
35.	<p>(a) <u>Procedure</u></p> <p>Take a small aluminium rod AB and using two connecting wires suspend it horizontally from a stand.</p> <p>Place a strong horse-shoe magnet in such a way that the rod lies between the two poles perpendicularly.</p> <p>Connect the aluminium rod in series with a battery and a key. Now pass a current through the aluminium rod from one end to another.</p> <p style="text-align: center;">/</p> <p>(Procedure can also be explained with a <u>Diagram</u>)</p>  <p><u>Observation</u></p> <p>It is observed that the rod is displaced on passing current through it.</p> <p>(b) Magnetic field will be vertically downwards.</p>	1 1	 3

<p>36.</p>	<p>(a)</p>  <p>(any two cases)</p> <p>/</p> <ul style="list-style-type: none"> • Magnetic field pattern for a current carrying straight conductor- concentric circles. • Magnetic field pattern for a current carrying solenoid- magnetic field lines similar to that of a bar magnet . • Magnetic field pattern for a current carrying circular loop- a pair of concentric circles with parallel straight lines at the centre. <p>(any two cases)</p> <p>(b)</p> <ul style="list-style-type: none"> • At X. • Magnetic field decreases as the distance from the conducting wire increases. 	<p>1+1</p> <p>1/2</p> <p>1/2</p>	<p>3</p>
<p>37.</p>	<p>(a) The phenomenon of scattering of light by colloidal particles which makes the path of light visible is called Tyndall effect.</p> <p>(b) Scattered light may appear white. / Scattered light may appear of a colour of longer wavelength.</p> <p>(c) Red colour is least scattered by fog or smoke, so it can be seen in the same colour from a large distance.</p>	<p>1</p> <p>1</p> <p>1</p>	<p>3</p>
<p>38.</p>	<p>(a)</p> <ul style="list-style-type: none"> • Position – Image will form at 40 cm / 2F / C • Nature – Real and inverted <p>/</p> <p>Alternate answer $f = +20 \text{ cm}, u = -40 \text{ cm}$ Using Lens Formula $\frac{1}{v} - \frac{1}{u} = \frac{1}{f}$ $\frac{1}{v} - \frac{1}{-40} = \frac{1}{20}$ $\frac{1}{v} = \frac{1}{20} + \frac{1}{-40}$</p> <p>$v = +40 \text{ cm}$ Position – 40 cm on the other side of the lens Nature – Real and inverted</p>	<p>1/2</p> <p>1/2</p>	

	<p>(b)</p>  <p>(c) (i)</p> $f_1 = 30 \text{ cm} = 0.3 \text{ m}, \quad f_2 = -15 \text{ cm} = -0.15 \text{ m}$ $P = \frac{1}{f}$ $P_1 = \frac{+1}{0.3} \text{ D}; \quad P_2 = \frac{-1}{0.15} \text{ D}$ <p>Equivalent power, $P = P_1 + P_2$</p> $P = -3.33 \text{ D}$ <p>Equivalent focal length, $f = \frac{1}{P}$</p> $f = \frac{-1}{3.33} = -0.3 \text{ m} = -30 \text{ cm}$ <p style="text-align: center;">OR</p> <p>(c) (ii)</p> <ul style="list-style-type: none"> • Combination Lens will behave like convex lens • $f_1 = -2 \text{ m}, f_2 = 1.5 \text{ m}$ $P = \frac{1}{f}$ $P_1 = \frac{-1}{2} \text{ D}, P_2 = \frac{+1}{1.5} \text{ D}$ $P = P_1 + P_2$ $P = \frac{1}{6}$ $f = +6 \text{ m}$ <p>The focal length of combination is positive.</p> <p style="text-align: center;">/</p> <p style="text-align: center;">Alternate answer</p> <ul style="list-style-type: none"> • Combination Lens will behave as a convex lens. • Convex lens – Less ‘f’, More ‘P₁’ Concave lens- More ‘f’, Less ‘P₂’ Combined Power = P₁+P₂, which will be positive. 	<p>1</p> <p>$\frac{1}{2}$</p> <p>$\frac{1}{2}$</p> <p>$\frac{1}{2}$</p> <p>1</p> <p>1</p>	4
<p>39.</p>	<p>(a)</p> <p>(i) (I) $R_s = 2 + 2 = 4 \Omega$</p> $\frac{1}{R'} = \frac{1}{4} + \frac{1}{4} = \frac{2}{4}$ $R' = 2 \Omega$ $R'' = R' + 3 \Omega$ $R'' = 2 \Omega + 3 \Omega$ $R'' = 5 \Omega$	<p>$\frac{1}{2}$</p> <p>$\frac{1}{2}$</p>	

<p>(II) $I = \frac{V}{R}$ $I = \frac{10}{5}$ $I = 2A$</p> <p>(III) $V = IR$ $V = 2 \times 3$ $V = 6V$</p>	<p>$\frac{1}{2}$</p> <p>$\frac{1}{2}$</p> <p>1</p>	
<p>(ii) As the two bulbs are connected in parallel, the voltage across each of them will be the same. Let I_1 be the current drawn by 100 W bulb: $I_1 = \frac{P}{V} = \frac{100}{220} = \frac{5}{11} A$ Let I_2 be the current drawn by 60 W bulb: $I_2 = \frac{P}{V} = \frac{60}{220} = \frac{3}{11} A$ \therefore Current drawn from line $I = I_1 + I_2$ $I = \frac{5}{11} + \frac{3}{11} = \frac{8}{11} A = 0.727 A \approx 0.73 A$ Hence, 0.73 A current flows through the line.</p>	<p>$\frac{1}{2}$</p> <p>$\frac{1}{2}$</p> <p>1</p>	
OR		
<p>(b)</p>		
<p>(i)</p> <ul style="list-style-type: none"> The Potential difference V across the ends of given metallic wire in an electric circuit is directly proportional to current I flowing through it, provided its temperature remains the same. 	<p>1</p>	
	<p>1</p>	
<ul style="list-style-type: none"> Slope = $\frac{BC}{AC}$ $= \frac{V}{I}$ Thus, Slope = R 	<p>1</p>	
<p>(ii) Consider three resistors R_1, R_2, R_3 connected in series. Let V be the potential difference across the resistors and current I flows through the circuit</p>		



Potential difference across R_1, R_2, R_3 is

$$V_1 = IR_1, V_2 = IR_2, V_3 = IR_3$$

If R_S is equivalent resistance in series combination and I is the current through the circuit then

$$V = I R_S$$

The total potential difference,

$$V = V_1 + V_2 + V_3$$

$$I R_S = IR_1 + IR_2 + IR_3$$

$$R_S = R_1 + R_2 + R_3$$

$\frac{1}{2}$

$\frac{1}{2}$

$\frac{1}{2}$

$\frac{1}{2}$

5